"cutting edge science for the classroom" (88) 228-417

MSDS No .: GG0050 Revision Date: May 1, 2010 Approved by: James A. Bertsch

1 = Slight

3 = Serious

Fire

Reactivity

0

0

MSDS No.: GG00	50			
Section 1	Chemical Product and Company Information			
Product	GERMANIUM			
Synonyms	Germanium Metal Powder			
CHEMTREC 2	4 Hour Emergency Phone Number (800) 424-9300			
Section 2	Hazards Identification			
Emergency Ov	verview	0 = Minimal	Health	0

### WARNING!

Section 3

Germanium

MAY BE HARMFUL IF SWALLOWED, INHALED OR ABSORBED THROUGH 2 = Moderate SKIN, MAY CAUSE IRRITATION. Keep container in a well-ventilated pla Use with adequate ventilation. Avoid of Wash thoroughly after handling, Targe

Il-ventilated place. D			4 = Severe	Contact	0
tilation, Avoid contac handling, Target orga				HMIS	*
Composition / Info	ormation on Ingred	lients			
lame	CAS #	%	TLV Unit	8	
	7440-56-4	100%	None established		

			(ACGIH 2001)
		_	
Section A	Firet Aid Moseuroe		

### First Aid Measu

**Chemical Name** 

INGESTION: Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

#### Section 5 **Fire Fighting Measures**

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Use water spray to keep fire-exposed containers cool, Avoid formation of dust. Dust dispersed in air becomes explosive when exposed to ignition source.

Extinguishing Media: Carbon dioxide, dry chemical, water spray, alcohol foam.	0 = Minimal
Flash Point: N/A	1 = Slight 2 = Moderate
Autoignition temperature: N/A	3 = Serious 4 = Severe
Explosion Limits: Lower: N/A Upper: N/A	Nonel
Section 6 Accidental Release Measures	and a second second

Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Contain spillage, then collect with an electrically protected vacuum cleaner or by wet-sweeping and place in suitable container for disposal according to local regulations. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

(2008 EMERGENCY RESPONSE GUIDEBOOK, (PHH50-ERG2008), GUIDE PAGE NO. 170)

#### Section 7 Handling & Storage

FLAMMABLE STORAGE CODE RED

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eves, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from sources of ignition - No smoking. Use spark-proof tools.

### Section 8 **Exposure Controls / Personal Protection**

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

#### Section 9 **Physical & Chemical Properties**

Physical state: Solid. Appearance: Grey powder. Odor: No odor.	Boiling point: 2830°C (5126°F) Freezing / Melting point: 937°C (1719°F) Decomposition temperature: N/A Solubility: Insoluble.
pH: N/A Vapor pressure (mm Hg): N/A Vapor Density (Air = 1): 5.35 g/ml @ 25°C (77°F) Evaporation rate ( = 1): N/A Viscosity: N/A	Specific gravity (H <sub>2</sub> O = 1): 5.35 g/cm <sup>3</sup> Percent volatile (%): 100% Molecular formula: Ge Molecular weight: 72.59
Section 10 Stability & Reactivity	
Chemical stability: Stable	Hazardous polymerization: Will not occur.

Incompatibilities with other materials: Strong oxidizers, halogens.

Hazardous decomposition products: Germanium oxides.

#### Section 11 **Toxicological Information**

Effects of overexposure: May cause skin imitation. May be harmful if absorbed through the skin. May cause eye imitation. Material may be irritating to mucous membranes and upper respiratory tract. May be harmful if inhaled. May be harmful if swallowed. To the best of our knowledge, the chemical, physical and toxicological properties have not been thoroughly investigated.

RTECS No.: LY5200000 ORL-RAT LD50: N/A IVN-MUS Id50: N/A

#### Section 12 **Ecological Information**

Do not flush into surface water or sanitary sewer system.

#### Section 13 **Disposal Considerations**

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14 Transport Information

UN/NA number: UN3089

Shipping name: Metal powders, flammable, n.o.s., (Germanium)

Hazard class: 4,1

Packing group: II

Exceptions: Limited quantity equal to or less than 1 Kg.

Section 15 **Regulatory Information** 

TSCA-listed, EINECS-listed (231-164-3), DSL-listed, RCRA code D001.

#### Section 16 **Additional Information**

2 = Moderate

3 = Serious

4 = Severe

Reactivity

HMIS \*

Contact

2

1

Inncivating Science<sup>TM</sup> by Aldon Corporation 221 Rockwater (Breat "cutting edge science for the classroom" (085) 226-6177

MSDS No.: MM0022 4414-6408 A414-6408 A414-6408 Approved by: James A. Bertsch

MSDS No.: MM00	22			
Section 1	Chemical Product and Company Information			
Product	MAGNESIUM METAL TURNINGS			
Synonyms	N/A			
CHEMTREC 2	4 Hour Emergency Phone Number (800) 424-9300			
Section 2	Hazards Identification			
Emergency Ov	verview	0 = Minimal	Health	0
DANGER! FL	AMMABLE SOLID!	1 = Slight	Fire	2

DANGEROUS WHEN WET. KEEP AWAY FROM IGNITION SOURCES. May be irritating to skin, eyes and respiratory system. Avoid looking at the intense white flame. Target organs: None known.

Section 3	Composition / Information on Ingredients				
Chen	nical Name	CAS	#	%	TLV Units
Magnesium		7439-95-4	99.8%	TWA: 10 n	ng/m3 (Mg oxide furnes) (ACGIH 2001)
Section 4	First Aid Measures	-			

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: DO NOT use water or foam to extinguish fire. DO NOT use carbon dioxide or halogenated extinguishing agents. Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Magnesium reacts with water and acids to release hydrogen. Avoid direct viewing of magnesium fires as eye injury may result, use fire glasses. Powders form explosive mixtures with air which may be ignited by a spark.

Extinguishing Media: Use approved Class D extinguisher or smother with dry sand, dry clay or dry ground limestone and dry graphite.

Flash Point: 636°C (1175°F)

Autoignition temperature: 510°C (950°F)

Explosion Limits: Lower: N/A Upper: N/A

Section 6 Accidental Release Measures

Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Use non-sparking tools. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water after material pickup is complete. Avoid runoff into storm sewers and ditches which lead to waterways.

(2008 EMERGENCY RESPONSE GUIDEBOOK, (PHH50-ERG2008), GUIDE PAGE NO. 138)

Section 7 Handling & Storage

FLAMMABLE STORAGE CODE RED

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts or fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources. Protect from water and moisture.

# Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

## Section 9 Physical & Chemical Properties

Odor: No odor. pH: N/A Vapor pressure (n Vapor Density (Al	ery gray, metal turnings. nm Hg): 1 mm @ 621°C	Boiling point: 1110°C (2030°F) Freezing / Melting point: 651°C (1202°F) Decomposition temperature: N/A Solubility: Negligible. Specific gravity (H <sub>2</sub> O = 1): 1.74 @ 20°C Percent volatile (%): 99.8% Molecular formula: Mg Molecular weight: 24.3	
Section 10	Stability & Reactivity		
Chemical stability: Stable		Hazardous polymerization: Will not occur.	

Conditions to avoid: Excessive temperatures, heat, sparks, open fiame and other sources of ignition.

Incompatibilities with other materials: Magnesium will react with water and acids to release hydrogen. Also hazardous with chlorine, bromine, iodine and oxidizing agents.

Hazardous decomposition products: Hydrogen.

### Section 11 Toxicological Information

Effects of overexposure: Exposure to magnesium metal or oxide dust should be a low health risk by inhalation and should be treated as a nuisance dust. Exposure to magnesium oxide fume subsequent to burning can result in metal fume fever. The temporary symptoms can include fever, chills, nausea, vomiting and muscular pain. Onset of symptoms occurs 4-12 hours after exposure. May cause burns and corneal abrasions to the eyes. Particles of magnesium embedded in the skin may produce lesions that resist healing.

ORL-RAT LD50: N/A IHL-RAT LC50: N/A SKN-RBT LD50: N/A

### Section 12 Ecological Information

Data not yet available.

### Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information	
UN/NA number:	UN1869	
Shipping name	Magnesium	
Hazard class:	4.1	
Packing group:	111	
Exceptions: Li	d Qty ≤ 5 Kg.	
Section 15	Regulatory Information	
TSCA-listed El	IECS-listed (221-104.6) PCPA code D001	

I SCA-listed, EINECS-listed (231-104-6), RCRA code DU01

Section 16 Additional Information



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MSDS No.: AA0375 Revision Date: February 25, 2010 Approved by: James A. Bertsch

MSDS No.: AA037		
Section 1	Chemical Product and Company Information	_
Product	ANTIMONY METAL LUMPS	
Synonyms	Antimony Regulus; Stibium	
CHEMTREC 2	24 Hour Emergency Phone Number (800) 424-9300	

# Section 2 Hazards Identification

**Emergency Overview** 

### DANGER!

HAZARDOU'S DUST. TOXIC AS FUME. MAY BE HARMFUL IF SWALLOWED. Avoid contact with skin, eyes and clothing. Use with adequate ventilation. Store in a cool, dry place. Wash thoroughly after handling. Target organs: Respiratory and cardiovascular system, liver, kidneys.

1 = Slight 2 = Moderate	Fire	1
3 = Serious	Reactivity	2
4 = Severe	Contact	2
	HMIS	*

3

0 = Minimal Health

Section 3	Composition / Info	Composition / Information on Ingredients				
Chem	ical Name	CAS#	%	TLV Units		
Antimony metal		7440-36-0	100%	TWA: 0.5 mg/m <sup>3</sup> (ACGIH 2001)		
Section 4	First Aid Measures					

INGESTION: Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Fire or excessive heat may produce hazardous decomposition products to be produced as dust or fume. Non-combustible solid in bulk form, but a moderate explosion hazard in the form of dust when exposed to flame. Use water spray to keep fire-exposed containers cool. Note: Stibine, which is extremely toxic, is formed when antimony is exposed to nascent (freshly formed) hydrogen.

Extinguishing Media: Sand, dry chemical, or CO2 should be used on surrounding fire. Do NOT use water on fire where molten metal is present.

Flash Point: Non-flammable.

Autoignition temperature: N/A

Explosion Limits: Lower: N/A Upper: N/A

Section 6 Accidental Release Measures

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Handling & Storage

Section 7

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TOXIC STORAGE CODE BLUE

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from acids or reactive substances.

### Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Boiling point: 1750°C (3182°F)
Freezing / Meiting point: 630°C (1166°F)
Decomposition temperature: N/A
Solubility: Insoluble in water.
Specific gravity (H <sub>2</sub> O = 1): 6.68
Percent volatile (%): N/A
Molecular formula: Sb
Molecular weight: 121,75
Hazardous polymerization: Will not occur.

Incompatibilities with other materials: Strong oxidizers, acids, halogenated acids [Note: Stibine is formed when antimony is exposed to nascent hydrogen.]

Hazardous decomposition products: Antimony oxides.

### Section 11 Toxicological Information

Effects of overexposure: INHALATION: Inflammation of mucous membranes of nose and throat. INGESTION: Stomatitis, vomiting, diarrhea. SKIN: Irritation and eczematous eruption on the skin. EYES: Irritation. Exercise appropriate procedures to minimize potential hazards.

RTECS #: CC4025000 UNKNOWN-MAN: LDLo: 15 mg/kg

# Section 12 Ecological Information

Data not yet available.

## Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information		
UN/NA number: Shipping name: Hazard class: M Packing group: Exceptions: N/	Not Regulated. VA N/A		
Section 15	Regulatory Information		
TSCA-listed, EIN	ECS-listed (231-146-5), DSL-listed.		

Section 16 Additional Information



Innovating Science TM by Addon Corporation 221 Ruchester Birett "cuttling edge science for the classroom" (2017 V 54 15-050 (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (2012) (201

MSDS No.: BB0130 Revision Date: March 2, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information	
Product	BISMUTH METAL LUMPS	
Synonyms	N/A	

CHEMITKEC 24 Hour Emergency Phone Humber (000) 424-3

# Section 2 Hazards Identification

**Emergency Overview** 

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LOW HAZARD FOR USUAL LABORATORY HANDLING. Bismuth metal is of relatively low toxicity and has liittle or no effect of intact skin or mucous membranes. The metal itself is not particularly hazardous on ingestion. Store in a cool, dry place. Wash thoroughly after handling. Target organs: None known.

= Minimal	Health	0
= Slight	Fire	0
= Moderate = Serious	Reactivity	1
= Severa	Contact	0

Section 3	Composition /	Information on Ingred	ients	
Chem	nical Name	CAS#	%	TLV Units
Bismuth metal		7440-69-9	100%	None established. (ACGIH 2001)

### Section 4 First Aid Measures

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Fire or excessive heat may produce hazardous decomposition products to be produced as dust or fume. Non-combustible solid in bulk form, but a moderate explosion hazard in the form of dust when exposed to flame. Use water spray to keep fire-exposed containers cool.

<b>Extinguishing Media:</b> Sand, dry chemical, or CO2 should be used on surrounding fire. Do NOT use water on fire where molten metal is present.	0 = Minimal	
Flash Point: Non-flammable.		$\wedge \wedge$
Autoignition temperature: N/A	2 = Moderate 3 = Serious 4 = Severe	$\vee \vee$
Explosion Limits: Lower: N/A Upper: N/A	4 - 304010	None listed.
Section 6 Accidental Release Measures		

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

Handling & Storage

Section 7

ae

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from acids or acid fumes.

### Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Odor: No odor. pH: N/A Vapor pressure (n Vapor Density (Ai	trous, silvery gray or reddish. nm Hg): 1 @ 1021°C	Boiling point: 1562°C (2487°F) Freezing / Melting point: 272°C (521°F) Decomposition temperature: N/A Solubility: Insoluble in water. Specific gravity (H <sub>2</sub> O = 1): 9.80 (20/4°C) Percent volatile (%): N/A Molecular formula: Bi Molecular weight: 208.98	
Section 10	Stability & Reactivity		
Chemical stability	: Stable	Hazardous polymerization: Will not occur.	

Conditions to avoid: Excessive temperatures and sources of ignition.

Incompatibilities with other materials: Strong oxidizers, acids, amonium nitrate plus powdered Bismuth may cause violent reaction.

Hazardous decomposition products: Emits toxic fumes on contact with acids. When heated, burns in air to form Bismuth trioxide.

### Section 11 Toxicological Information

Effects of overexposure: Bismuth metal is of relatively low toxicity and has liittle or no effect of intact skin or mucous membranes. The metal itself is not particularly hazardous on ingestion. EYES AND SKIN: Prolonged or repeated contact may cause irritation. Exercise appropriate procedures to minimize potential hazards.

ORAL-RAT: LD50: N/A

# Section 12 Ecological Information

Data not yet available.

# Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

 Section 14
 Transport Information

 UN/NA number:
 N/A

 Shipping name:
 Not Regulated.

 Hazard class:
 N/A

 Packing group:
 N/A

 Exceptions:
 N/A

 Section 15
 Regulatory Information

 TSCA-listed, EINECS-listed (231-177-4), RCRA code D001, DSL-listed.

Section 16 Additional Information

"cutting edge science for the classroom" (36) 221 Rodeneer Betell

MSDS No.: CC0014 Revision Date: February 19, 2010 Approved by: James A. Bertsch

Section 1 Chemical Product and Company Information		
Product	CADMIUM METAL	
Synonyms	N/A	

CHEMITKEC 24 Hour Emergency Phone Number (800) 424-93

# Section 2 Hazards Identification

**Emergency Overview** 

10000 No. 000000

## WARNING!

0 = Minimal 1 = Slight 2 = Moderate 3 = Serious 4 = Severe HMIS \*

HARMFUL IF INHALED AS FUMES. POISONOUS FUMES MAY BE FORMED ON HEATING. WARNING: This product contains a chemical known to the State of California to cause cancer. Abrasive to skin. Wash thoroughly after handling. Target organs: Lungs, kidneys.

Section 3	Composition / Inform	nation on Ingred	ients	
Chem	ical Name	CAS#	%	TLV Units
Cadmium metal		7440-43-9	100%	TWA: 0.01 mg/m³ (as dust) (ACGIH 2001)
Section 4	First Aid Measures			

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Molten metals produce fume, vapor and/or dust that may be toxic and/or a respiratory irritant. Metal reacts with oxidizing agents.

Extinguishing Media: Sand, dry chemical, or CO2 should be used on surrounding fire. Do NOT use water on fire where molten metal is present. Flash Point: >767°C (1412°F) Autoignition temperature: N/A Explosion Limits: Lower: N/A Upper: N/A	0 = Minimal 1 = Slight 2 = Moderate 3 = Serious 4 = Severe	NFPA
Section 6 Accidental Release Measures		iterite portea.

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Section 7 Handling & Storage

TOXIC STORAGE CODE BLUE

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible materials.

## Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

## Section 9 Physical & Chemical Properties

Physical state: Solid. Appearance: Silvery gray, lustrous metal. Odor: No odor. pH: N/A Vapor pressure (mm Hg): N/A Vapor Density (Air = 1): N/A Evaporation rate (Butyl acetate = 1): N/A		Boiling point: 767°C (1412°F) Freezing / Melting point: 320.9°C (610°F) Decomposition temperature: N/A Solubility: Insoluble. Specific gravity (H <sub>2</sub> O = 1): 8.642 @ 20°C Percent volatile (%): N/A Molecular formula: Cd	
Viscosity: N/A		Molecular weight: 112.40	
Section 10	Stability & Reactivity		
Chamical stability	Stoble	Herendous polymerization: Mill pet occur	

Chemical stability: Stable

Hazardous polymerization: Will not occur.

Conditions to avoid: Excessive temperatures above 610°C that may produce fumes.

Incompatibilities with other materials: Cadmium dust can react vigorously with oxidizing agents. Tarnishes in moist air. Contact with Hydrazoic acid may cause explosion. Avoid storage or use near acids or alkaline hydroxides.

Hazardous decomposition products: Heat treatment, welding or soldering of cadmium metal will produce toxic cadmium oxide fumes.

### Section 11 Toxicological Information

Effects of overexposure: WARNING: THIS PRODUCT CONTAINS A CHEMICAL KNOWN TO THE STATE OF CALIFORNIA TO CAUSE CANCER. Risk of cancer depends on level and duration of exposure. All cadmium compounds may cause cancer. A single exposure to excessive levels of cadmium fumes or dust can cause severe lung imitation, chest pain and edema which may be fatal. Lower exposure levels may cause dryness of throat, cough, headache, shortness of breath and vomiting. May cause skin irritation and discoloration. May cause mechanical irritation to the eyes. Exercise appropriate procedures to minimize potential hazards.

### RTECS #: EU9800000

Oral-rat LD50: 2330 mg/kg	Parenteral-rat LD50: 5 mg/kg	Intravenous-rabbit LD50: 2.5 mg/kg
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Section 12 Ecological Information

Data not yet available. Do not flush into surface water or sanitary sewer system.

## Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport	Information	
UN/NA number:	N/A		
Shipping name:	Not Regulated.		
Hazard class: N	VA		
Packing group:	N/A		
Exceptions: N/A	1		
Section 15	Regulatory	Information	

TSCA-listed, EINECS-listed (231-152-8), RCRA code D006, DSL-listed, Ca Prop 65-listed.

Section 16 Additional Information

Innovating Science<sup>TM</sup> by Addon Corporation 221 Rectasser Breat "cutting edge science for the classroom" (Aven, W 1 46146400 (Bis) 2246177

MSDS No.: MM0187 Revision Date: July 5, 2011 Approved by: James A. Bertsch

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4 = Sev

MSDS No.: MM01	87			
Section 1	Chemical Product and Company Information			
Product	MANGANESE METAL, CHIPS			
Synonyms	N/A			
CHEMTREC 2	4 Hour Emergency Phone Number (800) 424-9300			
Section 2	Hazards Identification			
Emergency Or	verview	0 = Minimal	Health	1
WARNING!		1 = Slight 2 = Moderate	Fire	0
HARMEULIE	SWALLOWED OR INHALED MAY CAUSE IRRITATION	Z = Moderate		

TO EYES AND SKIN. Avoid contact with skin, eyes and clothing. Wash thoroughly after

in the second se	Fire	0
ous	Reactivity	0
ere	Contact	0
	HMIS	ŧ.

Avoid contact with skin, eyes and clothing. Wash thoroughly after handling. Keep container tightly closed. Store in a cool, dry place. Target organs: Central nervous system.

Section 3	Composition / Infor	osition / Information on Ingredients			
Chem	nical Name	CAS#	%	TLV Units	
Manganese		7439-96-5	100%	TWA: 0.2 mg/m <sup>3</sup> [Manganese and inorganic compounds, as Mn] (ACGIH 2001)	
Section 4	First Aid Measures	1			

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild scap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Fire or excessive heat may produce hazardous decomposition products to be produced as dust or fume. Use water spray to keep fire-exposed containers cool.

Extinguishing Media: Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.	0 = Minimal	
Flash Point: Non-flammable.	1 = Slight 2 = Moderate	$\langle \rangle$
Autoignition temperature: N/A	3 = Serious 4 = Severe	$\times$
Explosion Limits: Lower: N/A Upper: N/A	4 - 500010	None listed.
Section 6 Accidental Release Measures		

Evacuate personnel to safe area. Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Section 7 Handling & Storage

G

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse. Storaee: Store in a cool, dry, well-ventilated area away from incompatible substances. Avoid exposure to water

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Avoid exposure to water and moisture.

Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Boiling point: 2095°C (3803°F)
Freezing / Melting point: 2271°C (1244°F)
Decomposition temperature: N/A
Solubility: Insoluble.
Specific gravity (H <sub>2</sub> O = 1): N/A
Percent volatile (%): N/A
Molecular formula: Mn
Molecular weight: 54.94
Hazardous polymerization: Will not occur.

Incompatibilities with other materials: Oxidizing agents, reducing agents, hydroxides, cyanides, carbonates. Hazardous decomposition products: Manganese oxides.

## Section 11 Toxicological Information

Effects of overexposure: Chronic manganese poisoning primarily involves the central nervous system. Early symptoms include languor, sleepiness and weakness in the legs. A stolid mask-like appearance of the face, emotional disturbances such as uncontrollable laughter and a spastic gait with tendency to fall while walking are findings in more advanced cases. High incidence of pneumonia has been found in workers exposed to the dust or fume of some manganese compounds. Men exposed to manganese dust have showed a decrease in fertility. To the best of our knowledge the chemical, physical and toxicological properties have not been thoroughly investigated. Specific data is not available. Exercise appropriate procedures to minimize potential hazards.

RTECS #: 009625000

Section 12	Ecological Information	
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EC50 - Daphnia magna (water flea) - 40 mg/l - 48 h

## Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information
UN/NA number	: N/A : Not Regulated.
Hazard class:	
Packing group:	
Exceptions: N	IA
Section 15	Regulatory Information
TSCA-listed, Ell	NECS-listed (231-105-1), DSL-listed, WHMIS Classification-D2A [as powder].

Section 16 Additional Information

Innovating Science<sup>TM</sup> by Aldon Corporation 221 Radvester Bleest "cutting edge science for the classroom" (888) 228-9177

MSDS No.: GG0225 Revision Date: February 22, 2010 Approved by: James A, Bertsch

MSDS No.: GG02	225	
Section 1	Chemical Product and Company Information	
Product	GRAPHITE POWDER	
Synonyms	Natural graphite; Carbon, Black Lead	
CHEMTREC 2	24 Hour Emergency Phone Number (800) 424-9300	
Section 2	Hazarde Identification	

Emergency Overview		0 = Minimal	Health	1
CAUTION!		1 = Slight	Fire	0
DO NOT BREATHE DUST.		2 = Moderate 3 = Serious	Reactivity	0
	e ventilation. Avoid contact with skin, eyes and clothing.	4 = Severe	Contact	1
Target organs: No	ne known.		HMIS	*

Section 3	Composition /	mposition / Information on Ingredients			
Chen	nical Name	CAS#	%	TLV Units	
Graphite powder		7782-42-5	100%	TWA: 2 mg/m <sup>3</sup> Respirable fraction (ACGIH 2001)	
Section 4	First Aid Measu	ures			

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild scap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. During a fire, irritating and highly toxic gases may be generated by thermal decomposition or combustion. Use water spray to keep fire-exposed containers cool. Graphite is a conductor of electricity. Use caution when handling in areas where where contact with eletrical circuitry is possible.

Extinguishing Med Flash Point: Non-		de, dry chemical, water spray, alcohol foam.	0 = Minimal 1 = Slight
Autoignition tempe			2 = Moderati 3 = Serious 4 = Severe
Explosion Limits:	Lower: N/A U	pper: N/A	4 - Severe
Section 6	Accidental R	Release Measures	

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

### Section 7 Handling & Storage

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources.

## Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

Section 9 Physical & Chemical Prop	perties
Physical state: Solid.	Boiling point: N/A
Appearance: Black powder. Odor: No odor.	Freezing / Melting point: Decomposes. Decomposition temperature: N/A
pH: N/A	Solubility in water: Insoluble.
Vapor pressure (mm Hg): N/A	Specific gravity (H <sub>2</sub> O = 1): 2.20-2.35 Percent volatile (%): N/A
Vapor Density (Air = 1): N/A Evaporation rate ( = 1): N/A	Molecular formula: C
Viscosity: N/A	Molecular weight: 12.01
Section 10 Stability & Reactivity	
Chemical stability: Stable	Hazardous polymerization: Will not occur.

Conditions to avoid: Stable under normal and fire conditions.

Incompatibilities with other materials: Strong oxidizers.

Hazardous decomposition products: Oxides of carbon.

## Section 11 Toxicological Information

Effects of overexposure: May cause skin imitation. May cause eye irritation. Material may be irritating to mucous membranes and upper respiratory tract. May be harmful if inhaled. May be harmful if swallowed. To the best of our knowledge, the chemical, physical and toxicological properties have not been thoroughly investigated.

RTECS No.: MD9659600 ORL-RAT LD50: N/A

# Section 12 Ecological Information

Data not yet available.

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## Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information		
UN/NA number: N/ Shipping name: N/ Hazard class: N/A Packing group: N/A Exceptions: N/A	ot Regulated.		
Section 15	Regulatory Information		
TSCA-listed, EINEC	S-listed (231-955-3), DSL-listed.		
Section 16	Additional Information		

Innovating Science<sup>TM</sup> by Aldon Corporation 221 Rochester Breat "cutting edge science for the classroom" (380) 226-6177

MSDS No.: CC0260 Revision Date: February 19, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information	
Product	CHARCOAL, WOOD POWDER	
Synonyms	N/A	

CHEMTREC 24 Hour Emergency Phone Number (800) 424-9300

## Section 2 Hazards Identification

**Emergency Overview** 

MSDS No . CCO260

### WARNING!

0 = Minimal 1 = Slight 2 = Moderate 3 = Serious 4 = Severe HMIS \*

PROVIDED FOR EXHAUSTING FUMES TO THE OUTSIDE. Toxic carbon monoxide released during burning. Use with adequate ventilation. Do not breathe dust. Avoid contact with skin, eyes and clothing. Target organs: None known

DO NOT USE FOR INDOOR HEATING OR COOKING UNLESS VENTILATION IS

Cher	nical Name	CAS#	%	TLV Units
Charcoal		7440-44-0	100%	TWA: 15 mg/m <sup>3</sup> total dust 8 hour (ACGIH 2001)
Section 4	First Aid Measu	res		

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

## Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. During a fire, imitating and highly toxic gases may be generated by thermal decomposition or combustion. Use water spray to keep fire-exposed containers cool. Dust dispersed in air becomes explosive when exposed to ignition source.

Extinguishing Media: Carbon dioxide, dry chemical, water spray, alcohol foam.	P
Flash Point: Combustible. Autoignition temperature: N/A	0 = Minimal 1 = Slight 2 = Moderate 3 = Serious
Explosion Limits: Lower: N/A Upper: N/A	4 = Severe No
Section 6 Accidental Release Measures	

Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Section 7 Handling & Storage

FLAMMABLE STORAGE CODE RED

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources.

## Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Physical state: Solid.		Boiling point: Decomposes. Freezing / Melting point: 3652-3697°C(6606-6687°F Decomposition temperature: 450°C (842°F) Solubility: Insoluble.						
Appearance: Black powder. Odor: No odor. pH: 6.0 - 9.0								
						Vapor pressure (	mm Hg): Negligible.	Specific gravity (H,O = 1): 1.8-2.1
						Vapor Density (Air = 1): N/A		Percent volatile (%): 100%
Evaporation rate	(= 1): N/A	Molecular formula: C						
Viscosity: N/A		Molecular weight: 12.01						
Section 10 Stability & Reactivity								
Chemical stability: Stable		Hazardous polymerization: Will not occur.						
Conditions to av	oid. Stable under normal and fire	conditions						

Conditions to avoid: Stable under normal and fire conditions

Incompatibilities with other materials: Strong oxidizers.

Hazardous decomposition products: Oxides of carbon.

# Section 11 Toxicological Information

Effects of overexposure: May cause skin imitation. May be harmful if absorbed through the skin. May cause eye irritation. Material may be irritating to mucous membranes and upper respiratory tract. May be harmful if inhaled. May be harmful if swallowed. To the best of our knowledge, the chemical, physical and toxicological properties have not been thoroughly investigated.

RTECS No.: FF5250100 ORL-RAT LD50: N/A IVN-MUS Id50: 440 mg/kg

## Section 12 Ecological Information

Data not yet available.

# Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information		
UN/NA number: Shipping name: Hazard class: 4 Packing group: Exceptions: N/	Charcoal .2 III		
Section 15	Regulatory Information		
TSCA-listed. EIN	IECS-listed (231-153-3)		
Section 16	Additional Information		

The information contained herein is furnished without warranty of any kind. Employers should use this information only as a supplement to other information gathered by them and must make independent determinations of suitability and completeness of information from all sources to assure proper use of these materials and the safety and health of employees. \* Hazardous Materials Industrial Standards.

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Innovating Science<sup>TM</sup> by Aldon Corporation 221 Rechester Brend "cutting edge science for the classroom" (880) 228-917

MSDS No.: CC0285 Revision Date: February 19, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information	
Product	CHROMIUM METAL	
Synonyms	Chromium; Chrome	

# Section 2 Hazards Identification

MCDC NA . COOOD

Target organs: Lungs, kidneys.		HMIS	*
equate ventilation. Store in a cool, dry place. Wash thoroughly after handling.	4 = Severe	Contact	0
Suspect cancer hazard. Avoid contact with skin, eyes and clothing. Use with ad-	3 = Serious	Reactivity	0
CAUTION! TOXIC BY INHALATION OF DUST OR FUME.	1 = Slight 2 = Moderate	Fire	0
Emergency Overview	0 = Minimal	Health	0

Section 3	Composition / Inform	ation on Ingred	ients	
Chen	nical Name	CAS #	%	TLV Units (ACGIH 2001)
Chromium metal		7440-47-3	100%	TWA: 0.5 mg/m <sup>3</sup> dust or fume
Section 4	First Aid Measures	1		

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

## Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Fire or excessive heat may produce hazardous decomposition products to be produced as dust or fume. Dusts may form fiammable and explosive mixtures in air. Use water spray to keep fire-exposed containers cool.

Extinguishing Media: Dry chemical, sand or carbon dioxide. Flash Point: Not flammable. Autoignition temperature: 400°C (752°F) as dust Explosion Limits: Lower: N/A Upper: N/A	0 = Minimal 1 = Sight 2 = Moderate 3 = Serious 4 = Severe
Section 6 Accidental Release Measures	PROVIDE LIFE-DROL

Evacuate personnel to safe area. Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

### Handling & Storage

Section 7

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources.

## Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Physical state: Solid. Appearance: Gray metal chunks or pieces. Odor: No odor. pH: N/A Vapor pressure (mm Hg): N/A Vapor Density (Air = 1): N/A Evaporation rate (Butyl acetate = 1): N/A Viscosity: N/A	Boiling point:       2200°C (3992°F)         Freezing / Melting point:       1830°C (3326°F)         Decomposition temperature:       N/A         Solubility in water:       Insoluble.         Specific gravity (H <sub>2</sub> O = 1):       7.20 @ 20°C         Percent volatile (%):       N/A         Molecular formula:       Cr         Molecular weight:       52.00
Section 10 Stability & Reactivity	
Chemical stability: Stable Conditions to avoid: Avoid generation of airborne dusts.	Hazardous polymerization: Will not occur. Excessive temperatures and heat.

Incompatibilities with other materials: Attacked by caustic alkalies and alkali carbonates, acids and strong oxidizers.

Hazardous decomposition products: Chromium fume.

## Section 11 Toxicological Information

Effects of overexposure: WARNING: THIS PRODUCT CONTAINS A CHEMICAL KNOWN TO THE STATE OF CALIFORNIA TO CAUSE CANCER. Risk of cancer depends on level and duration of exposure. All chromium compounds may cause cancer. Overexposure to dust may irritate eyes, nose and throat. Prolonged inhalation of dust may cause pulmonary disease. Chromium pieces or flake may contain sharp edges, therefore, caution is advised in handling. Exercise appropriate procedures to minimize potential hazards.

RTECS #: GB4200000 UNREPORTED ROUTE -RAT LD50: 27,500 µg/kg

Section 12	Ecological Information
Data not yet available.	
Section 13	Disposal Considerations
ply to empty container.	nes are intended for the disposal of catalog-size quantities only. Federal regulations may ap- State and/or local regulations may be different. Dispose of in accordance with all local, state a or contract with a licensed chemical disposal agency.
Section 14	Transport Information
UN/NA number: N/A Shipping name: Not Hazard class: N/A Packing group: N/A Exceptions: N/A	
Section 15	Regulatory Information
TSCA-listed, EINECS-	listed (231-157-5), RCRA code D007, DSL-listed, Ca Prop 65-listed.

Section 16 Additional Information

Innovating Science<sup>TM</sup> by Aldon Corporation 221 Rochester Bitratt "cutting edge science for the classroom" (306) 226-6177

MSDS No.: CC0345 Revision Date: February 19, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information	
Product	COBALT METAL	
Synonyms	Cobalt	

CHEMTREC 24 Hour Emergency Phone Number (800) 424-9300

## Section 2 Hazards Identification

**Emergency Overview** 

MODO NI- . 000045

### WARNING!

Health	1
Fire	0
Reactivity	0
Contact	1
	Reactivity

MAY BE HARMFUL IF SWALLOWED OR INHALED AS DUST OR FUME. WARN-ING: This product contains a chemical known to the State of California to cause cancer. Harmful fumes may be formed on heating. Use with adequate ventilation. Wash thoroughly after handling. Target organs: Lungs.

Section 3	Composition / Inform	ation on ingred	ients	
Cherr	nical Name	CAS#	%	TLV Units
Cobalt metal		7440-48-4	100%	TWA: 0.02 mg/m <sup>3</sup> (ACGIH 2001)
Section 4	First Aid Measures	1		

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If imitation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Molten metals produce fume, vapor and/or dust that may be toxic and/or a respiratory irritant. Metal reacts with oxidizing agents. Fine powder forms flammable and explosive mixtures in air. Material in powder form capable of creating a dust explosion.

Extinguishing Media: Sand, dry chemical, or CO2 should be used on surrounding fire. Do NOT use water on fire where molten metal is present. Flash Point: N/A Autoignition temperature: >200°C (392°F) Explosion Limits: Lower: N/A Upper: N/A	0 = Minimal 1 = Slight 2 = Moderate 3 = Serious 4 = Severe	NFPA
Section 6 Accidental Release Measures		

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Section 7 Handling & Storage

GEN

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible materials.

### Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Physical state: Solid.	Boiling point: 2723°C (3550°F)
Appearance: Gray, magnetic pieces.	Freezing / Melting point: 1493°C (2720°F)
Odor: No odor.	Decomposition temperature: N/A
pH: N/A	Solubility: Insoluble.
Vapor pressure (mm Hg): Negligible.	Specific gravity (H <sub>2</sub> O = 1): 8.9 g/cc
Vapor Density (Air = 1): N/A	Percent volatile (%): N/A
Evaporation rate (= 1): N/A	Molecular formula: Co
Viscosity: N/A	Molecular weight: 58.93
Section 10 Stability & Reactivity	
Chemical stability: Stable	Hazardous polymerization: Will not occur.
Conditions to avoid: Excessive temperatures.	

Incompatibilities with other materials: Strong acids, strong oxidizers, acetylene, hydrazinium nitrate, ammonium nitrate, bromine pentafluoride.

Hazardous decomposition products: Metallic oxides.

### Section 11 Toxicological Information

Effects of overexposure: Suspect cancer hazard. INHALATION: Inhalation may cause pulmonary symptoms and irritation of respiratory tract. SKIN: Powder may cause dermatitis. Possibility of allergic skin reaction. EYES: Contact with eyes may cause serious irritation. IARC classified: Group 2B: Possibly carcinogenic to humans. Risk of cancer depends on level and duration of exposure. Exercise appropriate procedures to minimize potential hazards.

RTECS #: GF8750000

## Section 12 Ecological Information

Data not yet available. Do not flush into surface water or sanitary sewer system.

## Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport	Information
UN/NA number: Shipping name: Hazard class: N. Packing group: Exceptions: N/A	Not Regulated. /A N/A	
Section 15	Regulator	y Information

TSCA-listed, EINECS-listed (231-158-0), RCRA code D001, DSL-listed, Ca Prop 65-listed, WHMIS Classification-listed.

Section 16 Additional Information

Innovating Science M by Aldon Corporation 221 Restaurer Barnet "cutting edge science for the classroom" Awan, NY 14414-6408 (565) 228-6177

MSDS No .: CC0413 Revision Date: February 19, 2010 A none hu: James A Bertech

MSDS No.: CC04	3		100	roved by: James	A. Bertsch	
Section 1	Chemical Product	and Company Info	rmation			
Product	COPPER METAL PC	NDER				
Synonyms	N/A					
CHEMTREC 2	4 Hour Emergency Phone Nur	nber (800) 424-9300				
Section 2	Composition / Info	mation on Ingredi	ents			
Ch	emical Name	CAS #	%	TLV Units		_
Copper		7440-50-8 TV	100% T\ /A: 0.2 mg/ni <sup>3</sup> ft	WA: 1.0 mg/m <sup>3</sup> dusts me (ACGIH 2001)		
Section 3	Hazards Identificat	ion				
Emergency Ov	erview			0 = Minimal	Health	0
CAUTION!	THE METAL DUCT			1 = Slight	Fire	0
	THE METAL DUST.	ed as dust or fume.	May cause imit	2 = Moderate 3 = Serious	Reactivity	0
	eyes. Avoid contact with Nitri				Contact	0
oxides. Target	organs: Liver, kidneys.		-		HMIS	*

### Section 4 **First Aid Measures**

INGESTION: Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

#### Section 5 **Fire Fighting Measures**

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. During a fire, irritating and highly toxic gases may be generated by thermal decomposition or combustion. Use water spray to keep fire-exposed containers cool, Non-flammable and non-combustible solid, but air-bom dust may ignite. Do not use water to fight fires involving this material.

Explosion Limits; Lower: N/A Upper: N/A	Extinguishing Media: Use triclass, dry chemical fire extinguisher. Flash Point: Non-combustible. Autoignition temperature: N/A Explosion Limits; Lower: N/A Upper: N/A	0 = Minimal 1 = Slight 2 = Moderate 3 = Serious 4 = Severe None liste
-----------------------------------------	---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------	--------------------------------------------------------------------------------------

Use proper personal protective equipment as indicated in Section 8, Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with scap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

### Section 7 Handling & Storage

**GENERAL STORAGE CODE GREEN** 

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dust. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources.

Section 8 **Exposure Controls / Personal Protection** 

Engineering controls: Facilities storing or utilizing this material should be equipped with an evewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None needed in normal laboratory handling. If dusty conditions prevail, work in a ventilation hood or wear a NIOSH/MSHA-approved dust mask.

#### Section 9 **Physical & Chemical Properties**

Odor: No odor. pH: N/A Vapor pressure (mn Vapor Density (Air =	sh-brown, lustrous metal. n <b>Hg):</b> 1 mm @ 1628*C	Boiling point: 2595°C (4703°F) Freezing / Melting point: 1083°C (1981°F) Decomposition temperature: N/A Solubility: Insoluble. Specific gravity (H <sub>2</sub> O = 1): 8.92 @ 20°C Percent volatile (%): N/A Molecular formula: Cu Molecular weight: 63.55	
Section 10	Stability & Reactivity		
Chemical stability:	Stable	Hazardous polymerization: Will not occur.	

Conditions to avoid: Excessive temperatures, heat, sparks, open flame and other sources of ignition. Acids.

Incompatibilities with other materials: Strong acids, oxidizers, alkalies, bromates, chlorates, iodates, sodium azide, acetyline and halogens,

Hazardous decomposition products: Nitrogen oxide is reacted with nitric acid.

#### Section 11 **Toxicological Information**

Effects of overexposure: Inhalation of this material can cause intense sneezing, nausea, vomiting, weakness and metal fume fever. Ingestion of this material may cause moderate irritation to the stomach lining. If product gets into eves, comeal abrasions may occur. May cause irritation on contact with skin. Repeated or prolonged exposure may cause liver and kidney damage, with an increased risk with Wilson's disease.

ORL-RAT LD50: N/A RTECS #: GL5325000

#### Section 12 **Ecological Information**

Data not yet available.

#### Section 13 **Disposal Considerations**

These disposal guidelines are intended for the disposal of catalog-size guantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local state and federal regulations or contract with a licensed chemical disposal agency.

Section 14 Transport Information UN/NA number: N/A Shipping name: Not Regulated. Hazard class: N/A Packing group: N/A Exceptions: N/A Section 15 **Regulatory Information** 

TSCA-listed, EINECS-listed (231-159-6), RCRA code D001, Ca Prop 65-Not listed.

Additional Information Section 16

Innovating Science<sup>TM</sup> by Aldon Corporation 221 Recharger Street "cutting edge science for the classroom" (008) 228-6177

MSDS No .: **BB0003** Revision Date: March 2, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information	
Product	BARIUM METAL RODS	
Synonyms	N/A	

# KEC 24 Hour Emergency Phone Number (800) 424-9300

### Section 2 **Hazards Identification**

**Emergency Overview** 

MCDC No . DD0000

### DANGER! FLAMMABLE SOLID!

0 = Minimal	Health	1
1 = Slight	Fire	3
2 = Moderate 3 = Serious	Reactivity	2
4 = Severe	Contact	1

DANGEROUS WHEN WET, AVOID CONTACT WITH SKIN AND EYES. Store under argon or paraffin oil in airtight container. Protect from moisture and air, Avoid contact with skin, eyes and clothing. Store in a cool, dry place. Wash thoroughly after handling. Target organs: Central nervous system, kidneys.

Section 3	Composition / Inform	ation on Ingred	ients	
Chen	nical Name	CAS#	%	TLV Units
Banum metal		7440-39-3	100%	TWA: 0.5 mg/m <sup>3</sup> as Barium and solutecompounds as Ba (ACGIH 2001)
Section 4	First Aid Measures			

INGESTION: Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

#### Section 5 **Fire Fighting Measures**

General information: DO NOT use water or foam to extinguish fire. DO NOT use carbon dioxide or halogenated extinguishing agents. Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Reaction with water produces explosive hydrogen gas and enough heat to ignite gas/air mixture plus toxic and corrosive Barium hydroxide solution.

Extinguishing Media: Use approved Class D extinguisher or smother with dry sand, dry clay or dry ground limestone and dry graphite.

Flash Point: Flammable solid.

Autoignition temperature: N/A

Explosion Limits: Lower: N/A Upper: N/A

Section 6 **Accidental Release Measures** 

Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Use non-sparking tools. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water after material pickup is complete. Avoid runoff into storm sewers and ditches which lead to waterways.

(2008 EMERGENCY RESPONSE GUIDEBOOK, (PHH50-ERG2008), GUIDE PAGE NO. 138)

Section 7 Handling & Storage FLAMMABLE STORAGE CODE RED

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children, Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale fumes. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources. Avoid exposure to water and moisture.

#### Section 8 **Exposure Controls / Personal Protection**

Engineering controls: Facilities storing or utilizing this material should be equipped with an evewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

#### Section 9 **Physical & Chemical Properties**

Odor: No odor. pH: N/A	strous, silvery gray.	Boiling point: 1695°C (3085°F) Freezing / Melting point: 850°C (1562°F) Decomposition temperature: N/A Solubility in water: Reacts violently.	
Vapor pressure ( Vapor Density (A		Specific gravity (H <sub>2</sub> O = 1): 3.74 Percent volatile (%): N/A	
Evaporation rate Viscosity: N/A	(Butyl acetate = 1): N/A	Molecular formula: Ba Molecular weight: 137.34	
Section 10	Stability & Reactivity		
Chemical stabilit	v: Stable	Hazardous polymerization: Will not occur.	

Conditions to avoid: Excessive temperatures and other sources of ignition. Avoid contact with water.

Incompatibilities with other materials: Water, acids, oxidizers, chlorinated and fluorinated hydrocarbons such as CCI.

Hazardous decomposition products: Hydrogen (expolosive), barium hydroxide solution (caustic/toxic).

#### Section 11 **Toxicological Information**

Effects of overexposure: Caustic burns of skin, eves and mucous membranes. Comeal damage and blindness may result. Moderately toxic via oral exposure route. Exercise appropriate procedures to minimize potential hazards.

RTECS #: CQ8370000 ORAL-RAT: LD50: N/A

### Section 12 **Ecological Information**

Data not yet available.

### Section 13 **Disposal Considerations**

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information			
UN/NA number: Shipping name:				
Hazard class: 4				
Packing group: Exceptions: Ltd	II I Qty ≤ 0.5 Kg (1.1 lb)			
Section 15	Regulatory Information	1		
TSCA-listed, EIN	ECS-listed (231-149-1), DSL-listed.			
	A LINE ALL Francestory			

Section 16 Additional Information

The information contained herein is furnished without warranty of any kind. Employers should use this information only as a supplement to other information gathered by them and must make independent determinations of suitability and completeness of information from all sources to assure proper use of these materials and the safety and health of employees, \* Hazardous Materials Industrial Standards.

NFPA



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MSDS No : SS0159 Revision Date: November 5, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information
Product	SILVER METAL
Synonyms	Silver

CHEMTREC 24 Hour Emergency Phone Number (800) 424-9300

### Section 2 **Hazards Identification**

**Emergency Overview** 

11000 No . 000100

# CAUTION!

MAY BE HARMFUL IF SWALLOWED, INHALED OR ABSORBED THROUGH SKIN

Minimal	Health	1
Slight	Fire	2
Moderate Serious	Reactivity	1
Severe	Contact	2

Avoid contact with skin, eyes and clothing. Use with adequate ventilation. Store in a cool, dry place. Wash thoroughly after handling. Target organs: None known

Section 3	Composition / In	formation on Ingred	ients	
Chen	nical Name	CAS #	%	TLV Units
Silver metal		7440-22-4	100%	TWA:0.1 mg/m²(air) (ACGIH 2001)
Section 4	First Aid Measure	es l		

INGESTION: Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

#### Section 5 **Fire Fighting Measures**

General information: Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Molten metals produce fume, vapor and/or dust that may be toxic and/or a respiratory irritant. Metal reacts with oxidizing agents. Fine powder forms flammable and explosive mixtures in air. Material in powder form capable of creating a dust explosion.

Extinguishing Media: Sand, dry chemical, or CO2 should be used on surrounding fire.		NFPA
Do NOT use water on fire where molten metal is present.	0 = Minimal	$\wedge$
Flash Point: N/A	1 = Slight 2 = Moderate	$\wedge$
Autoignition temperature: N/A	3 = Serious 4 = Severe	$\searrow$
Explosion Limits: Lower: N/A Upper: N/A	4 - Severe	None liste
Contine C Assidental Polones Meanures		

Evacuate personnel to safe area. Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with scap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

Section 7 Handling & Storage

**GENERAL STORAGE CODE GREEN** 

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible substances.

#### Section 8 **Exposure Controls / Personal Protection**

Engineering controls: Facilities storing or utilizing this material should be equipped with an evewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

#### **Physical & Chemical Properties** Section 9

Physical state: Solid.		Boiling point: 2212°C (4014°F)
Appearance: Silvery gra	ay, powder.	Freezing / Melting point: 962°C (1764°F)
Odor: No odor.		Decomposition temperature: N/A
pH: N/A		Solubility: Insoluble.
Vapor pressure (mm Hg	): N/A	Specific gravity (H,O = 1): 10.5
Vapor Density (Air = 1):	N/A as solid.	Percent volatile (%): N/A
Evaporation rate (= 1):	N/A	Molecular formula: Ag
Viscosity: N/A		Molecular weight: 107.87
Section 10 S	tability & Reactivity	
Chemical stability: Stal	ble	Hazardous polymerization: Will not occur.

Chemical stability: Stable

Conditions to avoid: Excessive temperatures and heat.

Incompatibilities with other materials: Acetylene, ammonia and hydrogen peroxide, nitric acid, ethylene imine, chlorine trifluoride, inorganic and organic peroxides, peroxyformic acid, bromoazide, oxalic acid, 1-bromo-2-propyne, tartaric acid, permonosulfuric acid, sulfuric acid.

Hazardous decomposition products: At temperatures above melting point, silver oxide fume may be evolved.

#### Section 11 **Toxicological Information**

Effects of overexposure: Silver metal as dust or fume presents a health hazard. Generalized argyria may result from silver ingestion. Exercise appropriate procedures to minimize potential hazards.

RTECS #: VW3500000

Section 12	Ecological Information	
Section 12	Ecological Information	

Data not yet available.

### Section 13 **Disposal Considerations**

These disposal guidelines are intended for the disposal of catalog-size guantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport	Information
UN/NA number:	N/A	
Shipping name:	Not Regulated.	
Hazard class: N	/A	
Packing group:	N/A	
Exceptions: N/A		
Section 15	Regulatory	Information

TSCA-listed, EINECS-listed (231-131-3), DSL-listed, WHMIS-Uncontrolled product,

Section 16 Additional Information

Innovating Science<sup>TM</sup> by Adon Corporation 221 Rochester Binet "cutting edge science for the classroom" (Aron, WT 19414-4000 (Big) 2224-177

MSDS No.: CC0036 Revision Date: February 19, 2010 Approved by: James A. Bertsch

MSDS NO.: CCUU		
Section 1	Chemical Product and Company Information	
Product	CALCIUM METAL	
Synonyms	N/A	
CHEMTREC 2	24 Hour Emergency Phone Number (800) 424-9300	

CHEMIREC 24 Hour Emergency Phone Number (000) 424-53

# Section 2 Hazards Identification

### **Emergency Overview**

DANGER! FLAMMABLE SOLID! DANGEROUS WHEN WET!

0 = Minimal	Health	1
1 = Slight	Fire	2
t = Moderate	Reactivity	2
= Severe	Contact	1
- Devele	HMIS	

NFPA

3

2 = Moderate

3 = Serious

4 = Severe

Reacts violently and/or explosively with water, steam or moisture to liberate hydrogen. Can be explosive when exposed to heat or flame. Causes severe irritation to the eyes, skin and mucous membrane with possible burns. Target organs: None known.

Section 3	Composition / Inform	1	1	
Cherr	nical Name	CAS #	%	TLV Units
Calcium		7440-70-2	98% T	None established. (ACGIH 2001)
Section 4	First Aid Measures	1	1	

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If imitation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: DO NOT use water or foam to extinguish fire. DO NOT use carbon dioxide or halogenated extinguishing agents. Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Calcium reacts with water and acids to release hydrogen. Heat of reaction may ignite evolved hydrogen. Avoid direct viewing of calcium fires as eye injury may result, use fire glasses. Powders form explosive mixtures with air which may be ignited by a spark.

Extinguishing Media: Use approved Class D extinguisher or smother with dry sand,	
dry clay or dry ground limestone and dry graphite.	0 = Minima
Flash Point: N/A	1 = Slight

Autoignition temperature: N/A

Explosion Limits: Lower: N/A Upper: N/A

Section 6 Accidental Release Measures

Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Use non-sparking tools. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water after material pickup is complete. Avoid runoff into storm sewers and ditches which lead to waterways.

(2008 EMERGENCY RESPONSE GUIDEBOOK, (PHH50-ERG2008), GUIDE PAGE NO. 138)

Section 7 Handling & Storage

FLAMMABLE STORAGE CODE RED

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dust. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources. Avoid exposure to water and moisture.

### Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower. Personnel should wear safety glasses, goggles, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If misty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Physical state: Solid.	Boiling point: 1440°C (4755°F)		
Appearance: Silvery gray metal.	Freezing / Melting point: 842°C (1542°F)		
Odor: No odor.	Decomposition temperature: N/A		
pH: N/A	Solubility: Reacts violently with water.		
Vapor pressure (mm Hg): 10 mm Hg @ 983°C	Specific gravity (H,O = 1): 1.54		
Vapor Density (Air = 1): N/A	Percent volatile (%): 98%		
Evaporation rate (Butyl acetate = 1): N/A	Molecular formula: Ca		
Viscosity: N/A	Molecular weight: 40.08		
Section 10 Stability & Reactivity			
Chemical stability: Stable	Hazardous polymerization: Will not occur.		

Conditions to avoid: Excessive temperatures, heat, sparks, open flame and other sources of ignition. Avoid contact with air and water. Avoid dispersion of dust in air.

Incompatibilities with other materials: Acids, halogens, alkali, metals, hydroxides, carbonates. Calcium will react with water and acids to release hydrogen.

Hazardous decomposition products: Hydrogen, calcium hydroxide, calcium oxide.

### Section 11 Toxicological Information

Effects of overexposure: This product is severely initiating to the eyes and may cause burns. Irritating to the skin. If ingested, product may produce gastrointestinal tract imitation and disturbances. If inhaled, dusts may cause irritation of the nose, throat, and respiratory tract. If this material contact moisture in the eyes, on the skin, or in the respiratory tract, severe irritation and corrosion of tissue may result.

ORL-RAT LD50: N/A IHL-RAT LC50: N/A SKN-RBT LD50: N/A

Section 12 Ecological Information

Data not yet available.

## Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14 Transport Information UNNA number: UN1401 Shipping name: Calcium Hazard class: 4.3

Packing group: II

Exceptions: Ltd Qty ≤ (0.5 Kg) 1.1 lb.

Section 15 Regulatory Information

TSCA-listed, EINECS-listed (231-179-5), RCRA code D001, D0003

### Section 16 Additional Information

Innovating Science<sup>TM</sup> by Aldon Corporation 221 Rochweise Beneti "cutting edge science for the classroom" (986) 2204177

MSDS No.: AA0145 Revision Date: January 26, 2010 Approved by: James A. Bertsch

MSDS No.: AA014	5		
Section 1	Chemical Product and Company Information		
Product ALUMINUM METAL			
Synonyms N/A			
CHEMTREC 2	4 Hour Emergency Phone Number (800) 424-9300		
Section 2	Hazards Identification		
Emergency On	0 = Minimal	Health	0
CAUTION!	1 = Slight	Fire	1
INHALATION	AS FUME MAY CAUSE IRRITATION. 2 = Moderate	Reactivity	1

Store away from acids, alkalies and oxidizers. Target organs: None known.

Slight Moderate Serious	Fire	1	
	Reactivity	1	
	Reactivity		
Severe	Contact	0	
	HMIS	*	

NFPA

Section 3	Composition / Information on Ingredients			
Cher	nical Name	CAS#	%	TLV Units
Aluminum		7429-90-5	> 99.5%	TWA: 10 mg/m <sup>3</sup> (Al metal dust) (ACGIH 2001)
Section 4	First Aid Measur	es	1	

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild scap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Molten metals produce fume, vapor and/or dust that may be toxic and/or a respiratory irritant. Metal reacts with oxidizing agents. Reacts with some acids and caustic solutions to produce hydrogen. Molten aluminum may explode on contact with water. It may also react violently with rust, certain metal oxides (e.g. oxides of copper, iron and lead) and nitrates (e.g. ammonium nitrate and fertilizers containing ammonium nitrate).

 Extinguishing Media:
 Sand, dry chemical, or CO2 should be used on surrounding fire.

 Do NOT use water on fire where molten metal is present.
 0 = Minima

 Flash Point:
 N/A

Autolgnition temperature: N/A	2 = Modera 3 = Seriout
Explosion Limits: Lower: N/A Upper: N/A	4 = Severe
Section 6 Accidental Release Measures	

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Wet-sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water after material pickup is complete. Avoid runoff into storm sewers and ditches which lead to waterways. Section 7 Handling & Storage

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhate fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible substances.

## Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: Use a chemical fume hood and/or wear a NIOSH/MSHA-approved respirator.

Section 9 Physical & Chemical Propertie	26
Physical state: Solid. Appearance: Silvery gray, spherical metal pieces. Odor: No odor. pH: N/A Vapor pressure (mm Hg): N/A Vapor Density (Air = 1): 0.095 - 0.113 lb/in <sup>3</sup> Evaporation rate (Butyl acetate = 1): N/A Viscosity: N/A	Boiling point: N/A Freezing / Melting point: 660°C (1220°F) Decomposition temperature: N/A Solubility: Insoluble. Specific gravity (H <sub>2</sub> O = 1): N/A Percent volatile (%): N/A Molecular formula: Al Molecular weight: 26.98
Section 10 Stability & Reactivity	
Chemical stability: Stable Conditions to avoid: Excessive temperatures and heat.	Hazardous polymerization: Will not occur.

conditions to avoid. Excessive temperatures and neat.

Incompatibilities with other materials: Strong oxidizers, mineral acids, strong alkalies, halogenated hydrocarbons, and water.

Hazardous decomposition products: Reacts with water (in molten form), acids or alkalies to generate hydrogen gas.

## Section 11 Toxicological Information

Effects of overexposure: May cause irritation. It has been reported that chronic exposure has been suspected of causing lung injury. To the best of our knowledge, the chemical, physical and toxicological properties have not been thoroughly investigated. Specific data is not available. Exercise appropriate procedures to minimize potential hazards.

ORL-RAT LD50: N/A IHL-RAT LC50: N/A SKN-RBT LD50: N/A

## Section 12 Ecological Information

Data not yet available

# Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport	Information		
UN/NA number: Shipping name: Hazard class:	Not Regulated.			
Packing group: Exceptions: N	N/A			
Section 15	Regulator	y information		
TSCA-listed, EIN	ECS-listed (231-0	72-3)		

Section 16 Additional Information

Innovating Science TM by Aldon Corporation 211 Rochester Breast "cutting edge science for the classroom" (369) 226-117

MSDS No.: IX0209 Revision Date: March 1, 2010 Approved by: James A. Bertsch

MSDS No .: IX0209	9					
Section 1	Chemical Product and Company Information					
Product	IRON METAL FI	IRON METAL FILINGS, DEGREASED				
Synonyms	Iron Aggregate	Iron Aggregate				
CHEMTREC 2	4 Hour Emergency Phon	ne Number (800) 424-930	0			
Section 2	Composition	Composition / Information on Ingredients				
Ch	emical Name	CAS#	%	TLV Units	1	
Iron aggregate Contains: Iron Carbon Silicon Manganese Chromium		65997-19-5 1309-37-1 7440-44-0 7440-21-3 7439-96-5 7440-47-3	100% >90% <4.0% <3.0% <0.3-1.0% <0.0-0.2%	TWA: 5 mg/m <sup>3</sup> N/A TWA: 10 mg/m <sup>3</sup> TWA: 0.2 mg/m <sup>3</sup> TWA: 0.5 mg/m <sup>3</sup> (matter)	ed Cr IIII compounda) (ACGIII	H 20
Section 3	Hazards Ident	lification				
Emergency Ov	verview			0 = Minimal	Health	1
CAUTION!					Fire	0
Iron dust dispersed in air may constitute a fire and/or explosion hazard. Iron dust may cause irritation and/or inflammation of the skin, eyes.					Reactivity	1
mucous membranes and lungs. 4 = Severe					Contact	1
Target organs: None known. HMIS *				*		

### Section 4 First Aid Measures

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General Information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Use water spray to keep fire-exposed containers cool. A fire hazard in the form of a fine dust dispersed in air or by chemical reaction with strong oxidizers can be an explosion hazard, especially when heated.

Extinguishing Media: Use dry chemical, dry sand or graphite for extinguishing fire. Flash Point: N/A	0 = Minimal 1 = Slight
Autoignition temperature: N/A	2 = Moderate 3 = Serious 4 = Severe
Explosion Limits: Lower: N/A Upper: N/A	None liste
Section 6 Accidental Release Measures	

Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Section 7 Handling & Storage

**GENERAL STORAGE CODE GREEN** 

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dust. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources.

## Section 8 Exposure Controls / Personal Protection

Bharles ( A Alter 1 al Brandles

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None needed in normal laboratory handling. If dusty conditions prevail, work in a ventilation hood or wear a NIOSH/MSHA-approved dust mask.

Section 9	Physical & Chemical Prop	erues
Physical state: 3 Appearance: Gre Odor: No odor. pH: N/A Vapor pressure (i Vapor Density (A Evaporation rate Viscosity: N/A	wy particles. mm Hg): N/A	Boiling point: N/A Freezing / Melting point: 1508.49°C (2750°F) Decomposition temperature: N/A Solubility: Insoluble. Specific gravity (H <sub>2</sub> O = 1): 6.7 gm/cc Percent volatile (%): N/A Molecular formula: Mixture. Molecular weight: Mixture.
Section 10	Stability & Reactivity	
Chemical stability		Hazardous polymerization: Will not occur.

Incompatibilities with other materials: Strong oxidizers, organic acids, mineral acids, water,

Hazardous decomposition products: None.

## Section 11 Toxicological Information

Effects of overexposure: Iron dust is an eye, skin and mucous membrane irritant. May cause irritation and inflammation of the eyes and lungs. Exercise appropriate procedures to minimize potential hazards.

ORL-RAT LD50: 30 gm/kg

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### Section 12 Ecological Information

Data not yet available.

### Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport	Information		
UN/NA number:	N/A			
Shipping name:	Not Regulated.			
Hazard class: N	AVA			
Packing group:	N/A			
Exceptions: N/	A			
Section 15	Regulator	y Information		- 120
TOOA Pated FINI	500 Katad (004 0		0 . D 05 M	

TSCA-listed, EINECS-listed (231-096-4), RCRA code D001, Ca Prop 65-Not listed.

Section 16 Additional Information

Innovating Science<sup>TM</sup> by Akkon Corporation 221 Rochester Bitmet "cutting edge science for the classroom" (889) 228-6177

MSDS No .: 110085 Revision Date: February 11, 2010 Approved by: James A. Bertsch

0 = Minimal

Health

Reactivity

HMIS

Contact

Fire

3

0

0

1

Section 1	Chemical Product and Company Information	
Product	LEAD METAL	
Synonyms	N/A	

CHEMTREC 24 Hour Emergency Phone Number (800) 424-9300

#### Section 2 Hazards Identification

**Emergency Overview** 

MSDS No 110085

CAUTION!	1 = Slight
MAY BE HARMFUL OR FATAL IF SWALLOWED OR INHALED AS DUST OR	2 = Moderate
	3 = Serious
FUME. Suspect cancer hazard. Store away from incompatible materials.	4 = Severe
Target organs: Lungs, kidneys.	

Chemical Name	CAS#	%	TLV Units
Lead metal	7439-92-1	99+%	TWA: 0.05 mg/m³ (Lead and inorganic compounds as Pb) (ACGIH 2001)
Section 4 First Aid Meas	lines		

INGESTION: Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

#### Section 5 **Fire Fighting Measures**

General information: Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Molten metals produce fume, vapor and/or dust that may be toxic and/or a respiratory irritant. Metal reacts with oxidizing agents. Molten aluminum may explode on contact with water.

Extinguishing Media: Sand, dry chemical, or CO2 should be used on surrounding fire. Do NOT use water on fire where molten metal is present.		NFP
	0 = Minimal	$\wedge$
Flash Point: N/A	1 = Slight 2 = Moderate	$\sim$
Autoignition temperature: N/A	3 = Serious 4 = Severe	$\sim$
Explosion Limits: Lower: N/A Upper: N/A	4 = Severe	None lis
Section 6 Accidental Release Measures		

Use proper personal protective equipment as indicated in Section 8, Provide adequate ventilation, Recover for use if not contaminated. Wet-sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water after material pickup is complete. Avoid runoff into storm sewers and ditches which lead to waterways.

Section 7 Handling & Storage

**GENERAL STORAGE CODE GREEN** 

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible substances.

#### Section 8 **Exposure Controls / Personal Protection**

Engineering controls: Facilities storing or utilizing this material should be equipped with an evewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

Section 9 Physical & Chemical Propertie	5
Physical state: Solid. Appearance: Silvery gray, metallic shots. Odor: No odor. pH: N/A Vapor pressure (mm Hg): N/A Vapor Density (Air = 1): N/A Evaporation rate (Butyl acetate = 1): N/A Viscosity: N/A	Boiling point: 1753°C (3187°F) Freezing / Melting point: 327°C (621°F) Decomposition temperature: N/A Solubility: Insoluble. Specific gravity ( $H_2O = 1$ ): 11.34 (20/4°C) Percent volatile (%): N/A Molecular formula: Pb Molecular weight: 207.19
Section 10 Stability & Reactivity	
Chemical stability: Stable Conditions to avoid: Excessive temperatures and heat.	Hazardous polymerization: Will not occur.

Incompatibilities with other materials: Strong oxidizers, hydrogen peroxide, acids,

Hazardous decomposition products: Emits toxic fumes of lead.

#### Section 11 **Toxicological Information**

Effects of overexposure: WARNING: THIS PRODUCT CONTAINS A CHEMICAL KNOWN TO THE STATE OF CALIFORNIA TO CAUSE CANCER. Risk of cancer depends on level and duration of exposure. Sheets have sharps edges and may cause cuts or scratches. Not absorbed through skin. Contact with eyes may cause transient imitation. Ingestion may cause anorexia, vomiting, malaise, convulsions due to increased intracranial pressure. Inhalation of dust or fumes can cause lead poisoning. Exercise appropriate procedures to minimize potential hazards.

### RTECS #: OF7525000

Intraperitoneal-rabbit: Lowest published toxic dose: 50 mg/kg Oral-woman: Lowest published toxic dose: 450 mg/kg/6 year

#### Section 12 **Ecological Information**

Data not yet available.

#### Section 13 **Disposal Considerations**

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information
UN/NA number: Shipping name: Hazard class: M Packing group: Exceptions: N/	Not Regulated. N/A N/A
Section 15	Regulatory Information
TSCA-listed FIN	ECS-listed (231-100-4) RCRA code D008 DSI -listed Ca Prop 65-listed

Section 16 Additional Information

MSDS No.: TT0290 Revision Date: November 5, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information	
Product	TUNGSTEN METAL	
Synonyms	Wolfram	
CHEMTREC 2	24 Hour Emergency Phone Number (800) 424-9300	

# Section 2 Hazards Identification

**Emergency Overview** 

## WARNING!

	0 = Minimal	Health	1
	1 = Slight	Fire	2
	2 = Moderate 3 = Serious	Reactivity	0
	4 = Severe	Contact	1
а		HMIS	*

FLAME. Avoid contact with skin, eyes and clothing. Use with adequate ventilation. Store in a cool, dry place. Wash thoroughly after handling. Target organs: None known.

MODERATE FIRE HAZARD IN THE FORM OF DUST WHEN EXPOSED TO

Section 3	Composition / Informa	1 1	and the second	1
Chen	nical Name	CAS#	%	TLV Units (ACGIH 2001)
Tungsten metal		7440-33-7	100%	TWA: 5 mg/m³(as Tungsten metal and insoluble compounds) (ACGIH 2001)
Section 4	First Aid Measures			

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Non-flammable and non-combustible solid, but air-born dust may ignite. Mothen metals produce fume, vapor and/or dust that may be toxic and/or respiratory imitants. In contact with acids, emits hydrogen gas. Do not use water to fight fires involving this material.

Extinguishing Media: Use triclass, dry chemical fire extinguisher. Do not use water. Flash Point: Non-flammable.	0 = Minimal 1 = Slight 2 = Moderate	NF
Autoignition temperature: N/A	3 = Serious 4 = Severe	$\smallsetminus$
Explosion Limits: Lower: N/A Upper: N/A	4 - 387616	None
Section 6 Accidental Release Measures		

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Handling & Storage

Section 7

GENER

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dust. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible substances.

### Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None needed in normal laboratory handling. If dusty conditions prevail, work in a ventilation hood or wear a NIOSH/MSHA-approved dust mask.

### Section 9 Physical & Chemical Properties

Physical state: Solid.	Boiling point: 5660°C (10220°F)		
Appearance: Silvery, metallic powder.	Freezing / Melting point: 3410°C (6170°F) Decomposition temperature: N/A Solubility: Insoluble.		
Odor: No odor.			
pH: N/A			
Vapor pressure (mm Hg): N/A	Specific gravity (H <sub>2</sub> O = 1): 19.3 Percent volatile (%): N/A		
Vapor Density (Air = 1): N/A			
Evaporation rate ( = 1): N/A	Molecular formula: W		
Viscosity: N/A	Molecular weight: 183.85		
Section 10 Stability & Reactivity			
Chemical stability: Stable Conditions to avoid: Avoid dispersion of dust in air.	Hazardous polymerization: Will not occur.		

Incompatibilities with other materials: Ignites on contact with fluorine gas at room temperature. Contact with acids may generate flammable hydrogen gas. Avoid contact with strong oxidizers.

Hazardous decomposition products: None known.

### Section 11 Toxicological Information

Effects of overexposure: Material may be irritating to mucous membranes and upper respiratory tract. If product gets into eyes, comeal abrasions may occur. May cause abrasions on contact with skin. To the best of our knowledge, the chemical, physical and toxicological properties have not been thoroughly investigated. Specific data is not available. Exercise appropriate procedures to minimize potential hazards.

RTECS No: YO7175000 ORL-RAT LD50: N/A

## Section 12 Ecological Information

Data not yet available.

# Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information
UN/NA number: N// Shipping name: N/A Hazard class: N/A Packing group: N// Exceptions: N/A	ot Regulated.
Section 15	Regulatory Information

TSCA-listed, EINECS-listed (231-143-9), DSL-listed, WHMIS Classification-Uncontrolled product.

Section 16 Additional Information

Innovating Science<sup>TM</sup> by Aldon Corporation 221 Rochester Street "cutting edge science for the classroom" (369) 226-0177

MSDS No.: NN0100 Revision Date: November 19, 2010 Approved by: James A, Bertsch

> 3 0 0

METAL, SHOT	
	rgency Phone Number (800) 424-9300

# Section 2 Hazards Identification

Emergency Overview	0 = Minimal	Health	T
WARNING!	1 = Slight	Fire	t
HARMFUL AS DUST OR FUME. MAY CAUSE SKIN IRRITATION.	2 = Moderate 3 = Serious	Reactivity	1
WARNING: This product contains a chemical known to the State of	4 = Severe	Contact	Ī
California to cause cancer. Wash thoroughly after handling. Target organs: None known.		HMIS	

Section 3	Composition / Information on Ingredients			
Chen	nical Name	CAS#	%	TLV Units
Nickel metal		7440-02-0	100%	TWA: 1 mg/m <sup>3</sup> (ACGIH 2001)
Section 4	First Aid Measures	1 1		

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Molten metals produce fume, vapor and/or dust that may be toxic and/or a respiratory irritant. Metal reacts with oxidizing agents. Reacts with some acids and caustic solutions to produce hydrogen.

Extinguishing Media: Sand, dry chemical, or CO2 should be used on surrounding fire. Do NOT use water on fire where molten metal is present.

Flash Point: Flammable as dust.

Autoignition temperature: N/A

Explosion Limits: Lower: N/A Upper: N/A

Section 6 Accidental Release Measures

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Handling & Storage

Section 7

GENER

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes skin and clothing. Avoid ingestion. Do not inhale fumes. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from acids or reactive substances.

## Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Physical state:	Solid.	Boiling point: 2732°C (4950°F)
Appearance: Silv	very gray, metal foil.	Freezing / Melting point: 1452°C (2645°F)
Odor: No odor.		Decomposition temperature: N/A
pH: N/A		Solubility: Insoluble.
Vapor pressure (	mm Hg): 1 mm @ 1810°C	Specific gravity (H,O = 1): 8.90 @ 20°C
Vapor Density (A	ir = 1): N/A	Percent volatile (%): N/A
<b>Evaporation</b> rate	(Butyl acetate = 1): N/A	Molecular formula: Ni
Viscosity: N/A		Molecular weight: 58.71
Section 10	Stability & Reactivity	
Chemical stabilit	y: Stable bid: Excessive temperatures and heat.	Hazardous polymerization: Will not occur.

Incompatibilities with other materials: Ammonium nitrate, perchlorates, phosphorus, selenium, sulfur. Slowly attacked by dilute hydrochloric acid or sulfuric acid. Readily attacked by nitric acid.

Hazardous decomposition products: Reacts with mineral acids to generate hydrogen. Evolved hydrogen may become an explosion hazard. Heating nickel metal emits nickel dust or fumes.

### Section 11 Toxicological Information

Effects of overexposure: Risk of cancer depends on level and duration of exposure. IARC classified: Group 2B: Possibly carcinogenic to humans. May cause dermatitis in sensitive individuals. May cause sensitization by skin contact. To the best of our knowledge the chemical, physical and toxicological properties have not been thoroughly investigated. Specific data is not available. Exercise appropriate procedures to minimize potential hazards.

RTECS #: QR5950000 ORL-RAT Lowest published lethal dose: 500 mg/kg

# Section 12 Ecological Information

Data not yet available.

# Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

 Section 14
 Transport Information

 UN/NA number:
 N/A

 Shipping name:
 Not Regulated.

 Hazard class:
 N/A

 Packing group:
 N/A

 Exceptions:
 N/A

 Section 15
 Regulatory Information

TSCA-listed, EINECS-listed (231-111-4), RCRA code D001, D003, DSL-listed, Ca Prop 65-listed.

Section 16 Additional Information



Innovating Science<sup>TM</sup> by Aldon Corporation 221 Recharger Bined "cutting edge science for the classroom" (Aron, W1 146146000 (Bit) 228-0177

MSDS No.: SS0151 Revision Date: December 27, 2010 Approved by: James A. Bertsch

4 = Severe

Contact

HMIS

0

MSDS No.: SS015	1			
Section 1	Chemical Product and Company Information			
Product	SILICON METAL LUMPS			
Synonyms	Silicon Metal			
CHEMTREC 2	4 Hour Emergency Phone Number (800) 424-9300			
Section 2	Hazards Identification			
Emergency Ov	review	0 = Minimal	Health	0
CAUTION		1 = Slight	Fire	0
	DUST. DO NOT INHALE AS DUST OR FUME.	2 = Moderate 3 = Serious	Reactivity	0

IRRITANT AS DUST. DO NOT INHALE AS DUST OR FUME. Low hazard in the form of lumps. Wash thoroughly after handling. Target organs: None known.

Chemical Name	CAS #	%	TLV Units (ACGIH 2001)
Silicon metal lumps	7440-21-3	100%	TWA: 10 mg/m <sup>3</sup> for total dust containing no asbestos and less than 1% crystalline silica; STEL: 20 mg/m <sup>3</sup> as dust

Section 4 First Aid Measures

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. No associated hazards known at this time.

Extinguishing Media: Use any media suitable for extinguishing supporting fire. Flash Point: Not flammable. Autoignition temperature: N/A	0 = Minimal 1 = Slight 2 = Moderate 3 = Serious 4 = Severe
Explosion Limits: Lower: N/A Upper: N/A	None lister
Section 6 Accidental Release Measures	

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

### Section 7 Handling & Storage

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible substances.

## Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Physical state: Solid.	Boiling point: N/A
Appearance: Metallic silver lumps.	Freezing / Melting point: 1440°C (2594°F)
Odor: No odor.	Decomposition temperature: N/A
pH: N/A	Solubility: Complete.
Vapor pressure (mm Hg): N/A	Specific gravity (H <sub>2</sub> O = 1): 2.3
Vapor Density (Air = 1): N/A	Percent volatile (%): N/A
Evaporation rate (Butyl acetate = 1): N/A	Molecular formula: Si
Viscosity: N/A	Molecular weight: 28.09
Section 10 Stability & Reactivity	
Chemical stability: Stable	Hazardous polymerization: Will not occur.
Conditions to avoid: Avoid generation of airborne dust	
Incompatibilities with other materials: Acids and stro	na hanna

Hazardous decomposition products: None known.

### Section 11 Toxicological Information

Effects of overexposure: Inhalation of dust may cause irritation to lungs, eyes and mucous membranes. Low hazard in the form of lumps. Exercise appropriate procedures to minimize potential hazards.

RTECS #: VW0400000 ORAL-RAT LD50: 3160 mg/kg EYE-RABBIT: 3 mg/mild

Section 12	Ecological Information	

Data not yet available.

5

# Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport	Information		
UN/NA number: Shipping name: Hazard class: N Packing group: Exceptions: N/A	Not Regulated. /A N/A			
Section 15	Regulator	Information		
TOCA Estad EINE	CC listed (224.4	20 8) DCDA anda	Doot DSI listed	

TSCA-listed, EINECS-listed (231-130-8), RCRA code D001, DSL-listed.

Section 16 Additional Information

Innovating Science TM by Aldon Corporation 221 Rochester Bired "cutting edge science for the classroom" (201) 221 Rochester Bired "(201) 224-2177

MSDS No.: SS1085 Revision Date: August 8, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information	
Product	SULFUR	
Synonyms	Sulfur Flower, Sulfur Flour	

CHEMTREC 24 Hour Emergency Phone Number (800) 424-9300

### Section 2 Hazards Identification

**Emergency Overview** 

## WARNING! FLAMMABLE SOLID!

	0 = Minimal	Health	1
	1 = Slight	Fire	1
IN Iv	2 = Moderate 3 = Serious	Reactivity	0
	4 = Severe	Contact	1
g		HMIS	*

BURNING SULFUR EMITS HIGHLY TOXIC FUMES. SULFUR DUST SUSPENDED IN AIR IGNITES EASILY. MAY CAUSE ALLERGIC REACTION. Sulfur reacts explosively with strong oxidizing agents such as nitrates and chlorates. It will also react at moderate rates with alkalies. Keep away from heat, sparks and open flame. Avoid breathing dust. Use with adequate ventilation, Target organs: None known.

Section 3	3 Composition / Information on Ingredients			
Chen	nical Name	CAS#	%	TLV Units (ACGIH 2001)
Sulfur		7704-34-9	100%	TWA: 10 mg/m <sup>3</sup> (total dust) TWA: 5 mg/m <sup>3</sup> (respirable dust)
Section 4	First Aid Measures	1 1		1

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

## Section 5 Fire Fighting Measures

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. During a fire, imitating and highly toxic gases may be generated by thermal decomposition or combustion. Use water spray to keep fire-exposed containers cool. Fires involving a small amount of combustibles may be smothered by dry chemical. Easily ignitable, combustible solid. Dust or vapor fumes form explosive mixtures with air. Hazardous in contact with oxidizing materials, forming explosive mixtures.

Extinguishing Media: Use water fog. Flash Point: 207°C (405°F) Closed Cup	0 = Minimal 1 = Slight
Autoignition temperature: 248°C (478°F)	2 = Moderate 3 = Serious 4 = Severe
Explosion Limits: Lower: 3.3% Upper: 46.0%	
Section 6 Accidental Release Measures	

Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Avoid creating dust. Use non-sparking tools. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

(2008 EMERGENCY RESPONSE GUIDEBOOK, (PHH50-ERG2008), GUIDE # 133)

Section 7 Handling & Storage

FLAMMABLE STORAGE CODE RED

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from ignition sources.

Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Physical state: Soli	d.	Boiling point: 444°C (831°F)			
Appearance: Yellow	powder, crystals or broken lumps.	Freezing / Melting point: 116-121°C (242-251°F)			
Odor: Faint odor of n	otten eggs.	Decomposition temperature: N/A			
pH: N/A		Solubility: Insoluble.			
Vapor pressure (mm	Hg): 0 @ 138°C (280°F)	Specific gravity (H2O = 1): 2.04-2.07 @ 21°C (70°F)			
Vapor Density (Air =	1): N/A	Percent volatile (%): Negligible.			
Evaporation rate (Bu	ityl acetate = 1): N/A	Molecular formula: S			
Viscosity: N/A		Molecular weight: 32.06			
Section 10	Stability & Reactivity				
Chemical stability:	Stable	Hazardous polymerization: Will not occur.			
Conditions to avoid:	Excessive temperatures, heat, spa	arks, open flame and other sources of ignition.			

Incompatibilities with other materials: Reacts violently with strong oxidizing agents. Corrosive to copper and copper alloys. Damp sulfur will corrode steel.

Hazardous decomposition products: Sulfur dioxide.

## Section 11 Toxicological Information

Effects of overexposure: Sulfur is essentially non-toxic either through ingestion, inhalation or skin contact. There are, however, some individuals who may be allergic and must not be permitted in the area of exposure. Sulfur is an eye irritant, but if prompt treatment is applied no lasting injury will result.

ORL-RAT LD50: 5 g/kg EYE-HMN LC50: 8 ppm SKN-RBT LD50: N/A

## Section 12 Ecological Information

Aquatic toxicity: TLm96: 1000 ppm

## Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport	Information			
UN/NA number:	NA1350				
Shipping name:	Sulfur				
Hazard class: 9					
Packing group:	H				
Exceptions: No	exceptions.				
Section 15	Regulatory	Information			
TSCA-listed, EIN	ECS-listed (231-7	22-6)			

Section 16 Additional Information

Innovating Science<sup>TM</sup> by Addon Corporation 221 Rochwater Breef "cutting edge science for the classroom" Awa, W1 1411-4600 #693 228-8177

MSDS No.: TT0175 Revision Date: January 25, 2010 Approved by: James A. Bertsch

Section 1	Chemical Product and Company Information					
Product	TIN METAL					
Synonyms	N/A					
CHEMTREC 2	4 Hour Emergency Phone Nun	nber (800) 424-9300				
Section 2	Composition / Information on Ingredients					
Ch	emical Name	CAS#	%	TLV	Units	
Tin metal		7440-31-5	99.85%	TWA: 2 mg/m <sup>3</sup> (ACGIH 2		
Castion 2	Unerside International					
	Hazards Identificati	ion				
Emergency Ov		ion		0 = Mi		0
Emergency Ov CAUTION!	verview		USE CARE	1 = Sli	ght Fire	0
Emergency Ov CAUTION! HARMFUL IF I			USE CARE	1 = Sli	ght oderate rious	-
Emergency Ov CAUTION! HARMFUL IF I HANDLING.	verview	. SHARP EDGES.		1 = Sli WHEN 2 = Mo 3 = Se	ght Fire Reactivity were Contact	0 0 1
HANDLING.	verview INHALED AS DUST OR FUME action to skin and eyes. Avoid o	. SHARP EDGES.		1 = Sli WHEN 2 = Mo 3 = Se	ght coderate micus Fire Reactivity	0 0 1

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General Information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Non-flammable and non-combustible solid, but air-bom dust may ignite. Molten metals produce fume, vapor and/or dust that may be toxic and/or respiratory irritants. In contact with acids, emits hydrogen gas. This gas if flammable and can cause explosion. Do not use water to fight fires involving this material.

Extinguishing Media: Use triclass, dry chemical fire extinguisher. Do not use water. Flash Point: Non-flammable. Autoignition temperature: N/A Explosion Limits: Lower: N/A Upper: N/A	0 = Minimal 1 = Sight 2 = Moderate 3 = Serious 4 = Severs None listed.
Section 6 Accidental Release Measures	None listed.

Use proper personal protective equipment as indicated in Section 8. Provide adequate ventilation. Recover for use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways. Handling & Storage

Section 7

GENER

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dust. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible substances.

### Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None needed in normal laboratory handling. If dusty conditions prevail, work in a ventilation hood or wear a NIOSH/MSHA-approved dust mask.

Section 9	Physical & Chemical Prop	erties
Vapor Density (A	very, metallic. ( <b>mm Hg):</b> 1 mm @ 1492°C	Boiling point: 2270°C (4118°F) Freezing / Melting point: 231.9°C (449°F) Decomposition temperature: N/A Solubility: Insoluble. Specific gravity (H <sub>2</sub> O = 1): 7.31 @ 20°C Percent volatile (%): N/A Molecular formula: Sn Molecular weight: 118.69
Section 10	Stability & Reactivity	
Chemical stabilit	ty: Stable	Hazardous polymerization: Will not occur.

Conditions to avoid: Contact with acids.

Incompatibilities with other materials: Strong oxidizers, chlorine, cupric nitrate, potassium, sodium peroxide, sulfur, halogens, sulfuric chlorosulfuric and pyrosulfuric acid.

Hazardous decomposition products: Oxides of tin may be present. Reacts with acids, causing evolution of flammable hydrogen gas.

## Section 11 Toxicological Information

Effects of overexposure: Material may be irritating to mucous membranes and upper respiratory tract. If product gets into eyes, comeal abrasions may occur. May cause abrasions on contact with skin. To the best of our knowledge, the chemical, physical and toxicological properties have not been thoroughly investigated. Specific data is not available. Exercise appropriate procedures to minimize potential hazards.

RTECS No: XP7320000 ORL-RAT LD50: N/A

Section 12	Ecological Information	
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Data not yet available.

## Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information	
UN/NA number: Shipping name:	N/A Not Regulated.	
Hazard class: 1	N/A	
Packing group:	N/A	
Exceptions: N/	A	
Section 15	Regulatory Information	
TSCA-listed, EIN	ECS-listed (231-141-8), RCRA code D001	

Section 16 Additional Information

Innovating Science<sup>TM</sup> by Adon Corporation 221 Rectineter Reveal "cutting edge science for the classroom" Avon, NY 1411-5400 (2022) 221-177

MSDS No.: ZZ0015 Revision Date: January 30, 2010 Approved by: James A. Bertsch

4 = Severe

Contact

HMIS

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MSDS No.: ZZ001	5		
Section 1	Chemical Product and Company Information		
Product	ZINC METAL		
Synonyms	N/A		
CHEMTREC 2	4 Hour Emergency Phone Number (800) 424-9300		
Section 2	Hazards Identification		
Emergency Ov	erview 0 = Minimal	Health	0
CAUTION!	1 = Slight	Fire	1
USE CARE IN	HANDLING. ABRASIVE TO SKIN. 2 = Moderate 3 = Serious	Reactivity	2

Reacts with acids to liberate hydrogen gas, a flammable and explosive gas. Store away from heat, open flames, acids and acid fumes. In case of fire, smother with sand. Dust clouds may be explosive. Target organs: None known.

Section 3	Composition / Inform	ation on Ingred	ients			
Chen	nical Name	CAS#	%	TLV Units		
Zinc metal		7440-66-6	99%	None established. (ACGIH 2001)		
Section 4	First Aid Measures		_			

**INGESTION:** Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

### Section 5 Fire Fighting Measures

General information: Use water spray to keep fire-exposed containers cool. In fire conditions, wear a NIOSH/ MSHA-approved self-contained breathing apparatus and full protective gear. Metal reacts with oxidizing agents. Powders form explosive mixtures with air which may be ignited by a spark. Reacts with some acids and caustic solutions to produce hydrogen, an explosive condition may exist if this happens in confined spaces.

Extinguishing Media: Sand, dry chemical, or CO2 should be used on surrounding fire. Do NOT use water where molten metal is present.	0 = Minimal	NFPA
Flash Point: Non-flammable.	1 = Slight 2 = Moderate	$\wedge$
Autoignition temperature: N/A	3 = Serious	$\searrow$
Explosion Limits: Lower: N/A Upper: N/A	4 = Severe	None liste
Section 6 Accidental Release Measures		

Use proper personal protective equipment as indicated in Section 8. Remove all sources of ignition. Provide adequate ventilation. Recover for use if not contaminated. Use non-sparking tools. Wet-sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water after material pickup is complete. Avoid runoff into storm sewers and ditches which lead to waterways. Section 7 Handling & Storage

GENERAL STORAGE CODE GREEN

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Avoid exposure to water and moisture.

## Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

### Section 9 Physical & Chemical Properties

Physical state: Solid.		Boiling point: 907°C (1665°F)		
Appearance: Silvery gray, metallic.		Freezing / Melting point: 419°C (787°F)		
Odor: No odor.		Decomposition temperature: N/A		
pH: N/A		Solubility: Insoluble.		
Vapor pressure (m	m Hg): N/A	Specific gravity (H,O = 1): 7.12		
Vapor Density (Air = 1): N/A		Percent volatile (%): N/A		
Evaporation rate (Butyl acetate = 1): N/A		Molecular formula: Zn		
Viscosity: N/A		Molecular weight: 65.38		
Section 10	Stability & Reactivity			
Chemical stability: Stable		Hazardous polymerization: Will not occur.		

Conditions to avoid: Excessive temperatures and heat. Hydrogen may evolve when in contact with water or damp air.

Incompatibilities with other materials: Strong oxidizers, acids, alkalies, and water.

Hazardous decomposition products: Zinc oxides and zinc fumes. Reacts with water, acids or alkalies to generate hydrogen gas.

### Section 11 Toxicological Information

Effects of overexposure: When heated above 400°C (752°F), inhalation of the fumes may lead to metal fume fever. Mild to severe symptoms of chills and fever, profuse perspiration, weakness, nausea, vomiting and coughing can occur. Contact with eyes may cause irritation. Prolonged or repeated skin contact may cause skin irritation. Exercise appropriate procedures to minimize potential hazards.

ORL-RAT LD50: N/A IHL-RAT LC50: N/A SKN-RBT LD50: N/A

## Section 12 Ecological Information

Data not yet available.

### Section 13 Disposal Considerations

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information		
UN/NA number: Shipping name: Hazard class: I Packing group: Exceptions: N/	Not Regulated. N/A N/A		
Section 15	Regulatory Information		
TSCA-listed, EIN	ECS-listed (231-175-3), DSL-listed.		

Section 16 Additional Information